metal-organic papers

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Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ R factor = 0.023 wR factor = 0.068 Data-to-parameter ratio = 15.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Poly[[[(pyrazino[2,3-f][1,10]phenanthroline- $\kappa^2 N, N'$)cadmium(II)]- μ_3 -pyridine-2,5-dicarboxylato- $\kappa^3 O, O': N, O: O'': O''$] monohydrate]

In the title compound, $\{[Cd(C_7H_3NO_4)(C_{14}H_8N_4)]\cdot H_2O\}_n$, the Cd^{II} atom is seven-coordinated by three N atoms from one pyrazino[2,3-*f*][1,10]phenanthroline molecule and one pyridine-2,5-dicarboxylate (2,5-pdc) ligand, and four O atoms from three different 2,5-pdc anions in a distorted monocapped octahedral coordination geometry. The Cd^{II} atoms are bridged by the 2,5-pdc ligands, forming a layer.

Comment

1,10-Phenanthroline (phen) and its derivatives are important ligands for the formation of metal–organic complexes (Che, Su *et al.*, 2006; Che, Liu, Liu *et al.*, 2006). Pyrazino[2,3-*f*][1,10]-phenanthroline (Pyphen) as a phen derivative possesses an extended delocalized aromatic system. To date, only a few coordination architectures based on Pyphen have been described (Che Liu & Xu, 2006; Che *et al.*, 2006). Our studies continue with the pyridine-2,5-dicarboxylate dianion (2,5-H₂pdc) in {[Cd(2,5-pdc)(Pyphen)]·H₂O}_n, (I).



The Cd atom is coordinated by three N atoms from one Pyphen molecule and one 2,5-pdc ligand, and four O atoms from three different 2,5-pdc anions in a distorted monocapped octahedral coordination geometry (Fig. 1). Bond dimensions involving the metal atom are similar to those reported in the literature (Gao *et al.*, 2004). Two Cd atoms are bridged by the 2,5-pdc ligands through the carboxylate μ_2 -O atoms, forming a dimeric unit; the dimeric units are further connected by the 2,5-pdc ligands through the chelating carboxylate groups, generating a layer with (4,4)-grids (Fig. 2). Each corner of the (4,4)-grid is occupied by a binuclear subunit. O-H···O hydrogen bonds involvinge the uncoordinated water molecule

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Figure 1

The asymmetric unit of compound (I), together with symmetryequivalent atoms to complete the coordination of Cd1. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry codes: (i) 1 - x, 1 - y, 1 - z; (ii) 1 - x, $y - \frac{1}{2}, \frac{1}{2} - z$.]

and carboxylate O atoms stabilize the crystal structure (Table 2).

Experimental

The Pyphen ligand was synthesized according to the method of Che, Li *et al.* (2006). A methanol solution (28 ml) of pyrazino[2,3f][1,10]phenanthroline (118 mg, 0.5 mmol) was mixed with an aqueous solution (10 ml) of CdCl₂·2.5H₂O (114 mg, 0.5 mmol) and 2,5-H₂pdc (167 mg, 1 mmol). The mixture was heated to dissolve the reagents. The solution was filtered; crystals separated from the solution after several days in 40% yield.

Crystal data

$[Cd(C_7H_3NO_4)(C_{14}H_8N_4)] \cdot H_2O$	Z = 4
$M_r = 52/.76$	$D_x = 1./92 \text{ Mg m}^2$
Monoclinic, $P2_{1}/c$	Mo $K\alpha$ radiation
a = 10.754 (2) A	$\mu = 1.16 \text{ mm}^{-1}$
b = 10.631 (2) Å	T = 292 (2) K
c = 17.127 (3) Å	Block, colorless
$\beta = 92.64 \ (3)^{\circ}$	$0.30 \times 0.27 \times 0.23 \text{ mm}$
V = 1956.0 (6) Å ³	
Data collection	
Rigaku R-AXIS RAPID	18518 measured reflections
diffractometer	4451 independent reflections
() scans	3937 reflections with $I > 2\sigma(I)$

 $R_{\rm int}=0.026$

 $\theta_{\rm max} = 27.5^{\circ}$

Figure 2 View of the layer structure of (I). H atoms have been omitted.

Refinement

Refinement on F^2 w = $R[F^2 > 2\sigma(F^2)] = 0.023$ $wR(F^2) = 0.068$ $wR(F^2) = 0.068$ $\Delta \rho$ S = 1.12 (Δr) 4451 reflections $\Delta \rho$ 295 parameters $\Delta \rho$ H atoms treated by a mixture of
independent and constrained
refinement

$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.0334P)^2 \\ &+ 1.031P] \\ &where \ P = (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{max} = 0.001 \\ \Delta\rho_{max} = 0.50 \ e \ \text{\AA}^{-3} \\ \Delta\rho_{min} = -0.46 \ e \ \text{\AA}^{-3} \end{split}$$

Table 1 Selected geometric parameters (Å, °).

Cd1-O1 ⁱ	2.2625 (16)	Cd1-N1	2.3916 (19)
Cd1-O1	2.3658 (17)	Cd1-N2	2.380 (2)
Cd1-O3 ⁱⁱ	2.372 (2)	Cd1-N5	2.3742 (19)
Cd1-O4 ⁱⁱ	2.520 (2)		
O1 ⁱ -Cd1-O1	69.81 (7)	O1-Cd1-N5	67.76 (6)
O1 ⁱ -Cd1-O3 ⁱⁱ	119.88 (10)	O3 ⁱⁱ -Cd1-N1	92.34 (9)
$O1-Cd1-O3^{ii}$	88.73 (9)	O3 ⁱⁱ -Cd1-N2	154.57 (9)
O1 ⁱ -Cd1-O4 ⁱⁱ	81.48 (7)	O3 ⁱⁱ -Cd1-N5	73.73 (8)
O1-Cd1-O4 ⁱⁱ	108.62 (7)	$N1-Cd1-O4^{ii}$	88.44 (7)
O3 ⁱⁱ -Cd1-O4 ⁱⁱ	52.55 (8)	$N2-Cd1-O4^{ii}$	139.53 (7)
O1 ⁱ -Cd1-N1	126.53 (7)	N2-Cd1-N1	69.21 (7)
O1-Cd1-N1	158.96 (7)	N5-Cd1-N2	89.21 (7)
O1 ⁱ -Cd1-N2	85.52 (7)	N5-Cd1-N1	92.38 (7)
O1-Cd1-N2	102.35 (7)	$N5-Cd1-O4^{ii}$	126.24 (7)
O1 ⁱ -Cd1-N5	134.94 (6)		

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) -x + 1, $y - \frac{1}{2}$, $-z + \frac{1}{2}$.

Table 2

Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$ \begin{array}{c} \hline O1W - HW12 \cdots O2 \\ O1W - HW11 \cdots O4^{iii} \end{array} $	0.86 (3)	1.986 (15)	2.787 (3)	155 (3)
	0.874 (10)	1.96 (3)	2.798 (3)	162 (4)

Symmetry code: (iii) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms on C atoms were positioned geometrically and refined as riding atoms, with C-H = 0.93 Å and $U_{iso}(H) = 1.2 U_{eq}(C)$. The

Absorption correction: multi-scan

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.670, T_{\max} = 0.763$

water H atoms were located in a difference Fourier map, and were refined with a distance restraint of O-H = 0.85 (1) Å; $U_{iso}(H) = 1.2$ $U_{eq}(O)$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *PROCESS-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTLPlus* (Sheldrick, 1990); software used to prepare material for publication: *SHELXL97*.

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References

- Che, G.-B., Liu, C.-B., Cui, Y.-C. & Li, C.-B. (2006). Acta Cryst. E62, m1362m1364.
- Che, G.-B., Liu, C.-B. & Xu, Z.-L. (2006). Acta Cryst. E62, m1948-m1949.
- Che, G.-B., Li, W.-L., Kong, Z.-G., Su, Z.-S., Chu, B., Li, B., Zhang, Z.-Q., Hu, Z.-Z. & Chi, H.-J. (2006). Synth. Commun. 36, 2519–2524.
- Che, G.-B., Su, Z.-S., Li, W.-L., Chu, B., Li, M.-T., Hu, Z.-Z. & Zhang, Z.-Q. (2006). Appl. Phys. Lett. 89, 103511.
- Che, G.-B., Xu, Z.-L. & Liu, C.-B. (2006). Acta Cryst. E62, m1370-m1372.
- Gao, S., Liu, J.-W., Huo, L.-H., Zhao, H. & Zhao, J.-G. (2004). Acta Cryst. E60, m1875–m1877.
- Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
- Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
- Sheldrick, G. M. (1990). SHELXTL-Plus. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.